The Synthesis of Novel Polycyclic Heterocyclic Ring Systems *via* Photocyclization. 4 [1a]. Benzo[f]thieno[2',3':4,5]thieno[2,3-c]quinoline and Benzo[h]thieno[2',3':4,5]thieno[2,3-c]quinoline and the Total Assignment of Their ¹H- and ¹³C-NMR Spectra

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The synthesis of two novel polycyclic heterocyclic ring systems via photocyclization are reported. These are benzo[f]thieno[2',3':4,5]thieno[2,3-c]quinoline and benzo[h]thieno-[2',3':4,5]thieno[2,3-c]quinoline. The total assignment of their 1 H- and 13 C-nmr spectra was determined by utilizing two-dimensional nmr spectroscopic methods.

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We have been interested in the synthesis of novel polycyclic heterocyclic ring systems via photocyclization of the appropriate amides [1a] and the study of their spectroscopic properties [1b,3,4]. In a recent paper [5] we reported that the photocyclization of 3-chloro-N-(2-naphthyl)benzo[b]thiophene-2-carboxamide in benzene afforded only benzo[f][1]benzothieno[2,3-c]quinolin-8(7H)-one with no benzo[g][1]benzothieno[2,3-c]quinolin-7(6H)-one being detected (Equation 1). The structural confirmation was supported by a 1D-HOHAHA spectra [3]. In contrast, photocyclization of 3-chloro-N-(3-phenanthryl)benzo[b]thiophene-2-carboxamide led to an isomeric mixture of [1]benzothieno[2,3-c]naphtho[1,2-f]quinolin-6(5H)-one and [1]benzothieno[2,3-c]naphtho[2,1-g]quinolin-6(7H)one [1b] (Equation 2). As a continuation of our studies of novel heterocyclic ring systems we now report the extension of a photo-induced cyclization leading to two previously unknown heterocyclic ring systems, namely benzo[h]thieno[2',3':4,5]thieno[2,3-c]quinoline (6) and benzo[f]thieno[2',3':4,5]thieno[2,3-c]quinoline (11). Synthesis.

The requisite key intermediate, 3-chlorothieno[3,2-b]thiophene-2-carbonyl chloride (1), has been described by Wright, Jr. [6] and Gronowitz and Maltesson [7] by refluxing 3-(2-thienyl)acrylic acid and thionyl chloride in the presence of pyridine. Thus when carbonyl chloride 1 was allowed to react with 1-naphthylamine (2) in benzene solution, 3-chloro-N-(1-naphthyl)thieno[3,2-b]thiophene-2-carboxamide (3) was obtained in 87% yield. Irradiation of the amide 3 in benzene solution containing triethylamine with a 450 watt medium pressure mercury vapor lamp gave the only possible cyclization product, i.e. benzo[h]thieno[2',3':4,5]thieno[2,3-c]quinolin-6(5H)-one (4) in 69% yield. Chlorination of 4 was performed by refluxing in phosphorus oxychloride to yield 6-chlorobenzo[h]thieno[2',3':4,5]thieno[2,3-c]quinoline (5) in 47% yield. Catalytic dechlorination of 5 with 10% Pd-C in 2:1 benzene-methanol solution in the presence of potassium hydroxide provided the novel parent ring system benzo-[h]thieno[2',3':4,5]thieno[2,3-c]quinoline (6) in 49% yield after column chromatography (Scheme I).

Likewise, 3-chloro-N-(2-naphthyl)thieno[3,2-b]thiophene-2-carboxamide (8) was obtained in 79% yield upon treatment of 1 with 2-naphthylamine (7). Photocyclization of 8 provided only one of the two possible isomers, benzo[f]thieno[2',3':4,5]thieno[2,3-c]quinolin-8(7H)-one (9) in quantitative yield with no trace of benzo[g]thieno[2',3':4,5]thieno[2,3-c]quinolin-7(6H)-one (12) being formed. These results are consistent with our previous work [5]. The ^{1}H nmr shows no singlets beyond $\delta = 9.0$ which is anticipated for H-12 of 12 and its structurally analogous polycyclic aromatic hydrocarbons [8] and heterocycles [1b]. Chlorination of 9 was accomplished by refluxing in phosphorus oxychloride to afford 8-chloroben-

Scheme I

zo[f]thieno[2',3':4,5]thieno[2,3-c]quinoline (10) in 61% yield. Catalytic dechlorination of 10 with 10% Pd-C in 2:1 benzene-methanol in the presence of potassium hydroxide gave the unsubstituted novel heterocyclic ring system benzo[f]thieno[2',3':4,5]thieno[2,3-c]quinoline (11) in 59% yield (Scheme I).

NMR Spectroscopy

Spectral assignment (Table 1) of benzo[h]thieno-[2',3':4,5]thieno[2,3-c]quinoline (6) was accomplished us-

ing data from two-dimensional nmr spectra. Two protons further downfield than the rest, a singlet at $\delta = 9.39$ and a doublet centered at 9.37, were tentatively assigned as H-6 and H-4, respectively, based upon chemical shift, coupling considerations, and past experience [1b,3,5,9]. Through the utilization of an auto-correlated homonuclear nmr experiment (COSY, not shown) it is clear that all the members of the four spin system may be conveniently elucidated. Two of the two spin system show AM patterns and are yet to be assigned. Interestingly, although a five

bond epi-zig-zag coupling was well documented [10], the anticipated long-range coupling between H-4 and H-12 of 6 is not observable.

The connectivity of the protons with their respective carbon signals was established using heteronuclear chemical shift correlation with broadband homonuclear proton decoupling as described by Bax [11]. With all protons and protonated carbon atoms accounted for, the unequivocal

Table 1

1H- and 13C-NMR Chemical Shift Assignments and Observed
Proton-Carbon Multiple-Bond Correlations for Compound 6 in
Deuteriochloroform at 298°K at Observation Frequencies of 360.13 and
90.56 MHz, Respectively

Position	$\delta_{ m H}$	δ _C	Two-Bond Correlation	Three-Bond Correlation
1	7.98	127.90		H-3, H-12
2	7.73	127.60		H-4
2 3	7.78	127.29		H-1
4	9.37	124.67		H-2
4a		131.79		H-1, H-3, H-12
4b		142.31		H-4, H-6, H-11
6	9.39	144.03		•
6a		136.86	H-6	
7a		142.13		H-8, H-9
8	7.50	120.32	H-9	•
9	7.82	132.31	H-8	
10a		133.24		H-8, H-9
10b		134.88		H-6, H-11
10c		120.17	H-11	H-12
11	8.27	121.24		
12	8.02	128.33		H-1
12a		132.81	H-12	H-2, H-4, H-11

assignment of the quaternary carbons was made via a long-range heteronuclear chemical shift correlation, experiment [12] (Figure 1) to complete the total assignment of the ¹H- and ¹³C-nmr spectra of 6.

Several key long-range connectivities are critical to the assignment of the proton and carbon nmr spectra of 6. Once H-6 has been tentatively identified, it would in turn be expected to be long-range coupled to C-10b resonating at $\delta=134.88$ as well as C-4b at 142.31 on the basis of their chemical shifts. The verification of the C-4b assignment was provided by two additional three-bond correlations both to H-11, a two spin system, and H-4 in the adjacent four spin system. For C-10b, the assignment was confirmed by coupling to only one of the protons, H-11, from the two spin system. Confirmation of the assignment of C-6a was derived from the two-bond coupling between H-6 and C-6a at $\delta=136.86$, which would show no other possibility for a long-range response.

Here, H-12 would also be utilized as a key focal point in addition to H-6. Possible three-bond coupling pathways for H-12 would include coupling to C-10c, C-1, and C-4a, respectively. Long-range coupling responses to the other protonated and quaternary carbon signals were analyzed and assigned in a manner identical with that just described as above.

The assignment of the protons and carbon atoms of the terminal thiophene moiety was rather ambiguous. The signals at $\delta = 7.50$ and 7.82 are correlated with both of the signals at $\delta = 133.24$ and 142.13 (Figure 1). However, a close examination of the spectrum (Figure 2) revealed a response correlating the proton resonating at $\delta = 9.39$ with the carbon at $\delta = 133.24$, which was barely visible at the threshold of the contour plot. Differentiation from noise was provided by examination of the slice corresponding to this correlation, from the long-range heteronuclear experiment. This particular response may result from the four-bond coupling between H-6 and C-10a of 6. Although four-bond couplings are usually small and difficult to observe, a positive coupling across W path has been observed in pyrene [13], thienopyridines [13], and 7hydroxyfrullanolide [14,15]. Coupled with the knowledge that the chemical shift and that the three-bond heteronuclear couplings are generally substantially larger than their two-bond counterparts, the carbon atom resonating at δ = 133.24 may be assigned as C-10a and δ = 142.13 as C-7a. Accordingly, responses occurring at $\delta = 7.50$ and 7.82 may be immediately identified as H-8 and H-9. respectively. Thus it was possible to totally assign the proton and carbon spectra of 6 as collected in Table 1.

The spectral assignment of benzo[f]thieno[2',3':4,5]-thieno[2,3-c]quinoline (11) was analyzed and achieved in a fashion identical to that employed in the assignment of

Table 2

1H- and 13C-NMR Chemical Shift Assignments and Observed Proton-Carbon Multiple-Bond Correlations for Compound 11 in

Deuteriochloroform at 298°K at Observation Frequencies of 360.13 and

90.56 MHz, Respectively

Position	δ_{H}	δ _C	Two-Bond Correlation	Three-Bond Correlation
1	9.28	126.48		Н-3
2	7.70	125.48		H-4
3	7.70	127.26		H-1
4	7.98	128.30		H-2,H-5
4a		132.20	H-4, H-5	H-1, H-3, H-6
5	7.98	129.30	•	H-4
6	8.08	128.24		
6a		145.18		H-5,H-8
8	9.31	145.26		
8a		137.72	H-8	
9a		143.25	H-10	H-11
10	7.42	119.97	H-11	
11	7.60	130.01	H-10	
12a		135.19		H-10, H-11
12b		134.40		H-8
12c		120.97		H-1, H-6
12d		128.83		H-2, H-4, H-5

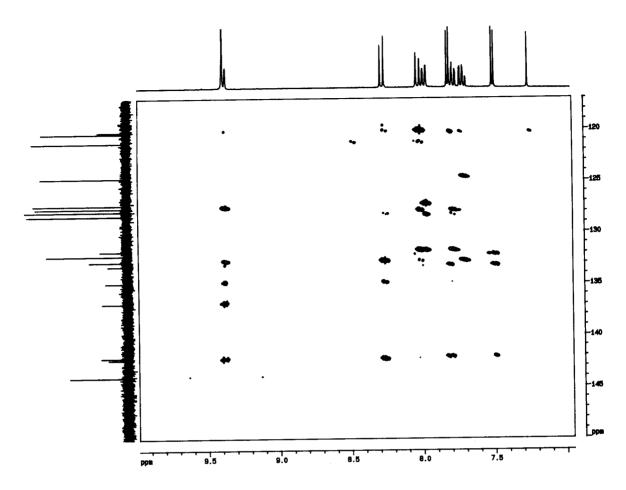


Figure 1. Long-range heteronuclear chemical shift correlation spectrum of 6 recorded in deuteriochloroform at observation frequencies of 360.13/90.56 MHz for ¹H and ¹³C, respectively.

the spectra of 6. Interestingly, in contrast to 6 the COSY spectrum (not shown) of 11 clearly demonstrated the anticipated five-bond epi-zig-zag coupling [10] between H-1 and H-5 (vide supra), linking the four spin system to the adjacent two spin system, and facilitated the assignment of benzo[f]quinoline moiety of 11. The assignment of the remaining resonances of the terminal thiophene moiety is rather ambiguous due to the lack of a four-bond coupling across a W path between H-8 and C-12a [16] (vide supra). Based upon the prior assignment of 6 (vide supra) there is little doubt that the unequivocal total assignment of 11 can be achieved (Table 2).

EXPERIMENTAL

Melting points were determined on a Thomas-Hoover melting point apparatus and are uncorrected. The ir spectra were recorded on a Beckman FT 1100 spectrometer as potassium bromide pellets and frequencies are expressed in cm⁻¹. The ¹H nmr spectra were obtained on a JEOL FX-90Q spectrometer in the solvent indicated with TMS as the internal standard and chemical shifts are reported in ppm (δ) and J values in Hz. Column chromatography was performed utilizing Aldrich silica gel, 70-230 mesh. Elemental analyses were performed by M-H-W Laboratories, Phoenix, Arizona.

Proton and carbon nmr spectra of final products were acquired using a Bruker AMX360 spectrometer operating at a proton frequency of 360.13 MHz and a carbon frequency of 90.56 MHz. The proton spectra were obtained using a 5 μ s (48.6°) pulse and 5 seconds between transients to insure accurate integrals. All two-dimensional experiments were acquired using a Bruker inverse geometry probe. Spectral width for all proton and proton correlation experiments was 992.06 Hz. A one-dimensional carbon spectrum was obtained with a sweep width of 15001.5 Hz with 2 seconds between transients. A phase

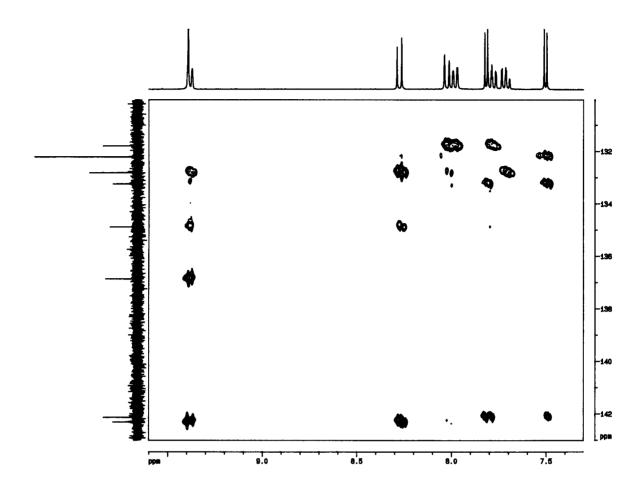


Figure 2. Expansion of the region from 131-143 ppm on F2 in Figure 1.

sensitive double quantum filtered COSY experiment was acquired [17] with 512 t_1 increments of 504 μ s used to encode a second dimension of 992.06 Hz. Proton-carbon correlation experiments were acquired using the proton-detected techniques of Bax [11,12]. Mixing times of 3.0 ms and 47.0 ms were used for direct and long-range methods, respectively. For the phase sensitive direct correlation technique, 256 t_1 increments of 156 μ s were used to encode a sweep width of 3205.13 Hz. For the long-range correlation experiment 256 t_1 increments of 312 μ s were used to encode a sweep width of 3205.13 Hz.

3-Chloro-N-(1-naphthyl)thieno[3,2-b]thiophene-2-carboxamide (3).

A mixture of 2.25 g (10 mmoles) of carbonyl chloride 1 [6,7], 1.43 g (10 mmoles) of 1-naphthylamine (2), and 40 ml of benzene was heated under reflux for 4 hours. After cooling the solid was collected by filtration and recrystallized from cyclohexane to give 2.99 g (8.70 mmoles, 87%) of 3 as colorless crystals, mp 160-162°; ir (potassium bromide): 3237 (NH stretching), 3091, 3073, 3050 (aromatic CH stretching), 1617

(C=O stretching); 1 H nmr (deuteriochloroform): δ 7.32 (d, 1H, $J_{5,6} = 5.4$ Hz, H-6), 7.43-8.27 (m, 8H, ArH including a doublet at δ 7.61 with $J_{5,6} = 5.4$ Hz assigned to H-5), 9.13 (br s, 1H, NH).

Anal. Calcd. for C₁₇H₁₀ClNOS₂: C, 59.38; H, 2.93; N, 4.07; S, 18.65. Found: C, 59.18; H, 3.07; N, 3.98; S, 18.49.

Benzo[h]thieno[2',3':4,5]thieno]2,3-c]quinolin-6(5H)-one (4).

A mixture of 0.5 g (1.45 mmoles) of 3, 0.159 g (1.45 mmoles) of triethylamine, and 500 ml of benzene was irradiated with a 450 watt Hanovia medium pressure mercury vapor lamp for 4 hours. A slow stream of air was passed through the solution during the course of the reaction. The solid was collected by filtration and washed with water to give 0.31 g (1.01 mmoles, 69%) of 4, mp >280°; ir (potassium bromide): 3160 (NH stretching), 3055 (aromatic CH stretching), 1653 (C=O stretching); $^1{\rm H}$ nmr (DMSO-d₆): 140° δ 7.55-8.11 (m, 7H, ArH), 8.92 (m, 1H, H-4). The compound was used for the next step without further purification because of low solubility.

6-Chlorobenzo[h]thieno[2',3':4,5]thieno[2,3-c]quinoline (5).

A mixture of 1.00 g (3.25 mmoles) of 4 and 25 ml of phosphorus oxychloride was heated at 110-120° for 4 hours. After cooling the mixture was poured into 350 ml of ice-water with caution. The solid was collected by filtration and recrystallized from benzene to afford 0.50 g (1.53 mmoles, 47%) of 5 as colorless crystals, mp 229-231°; ir (potassium bromide): 3065 (aromatic CH stretching); 1 H nmr (deuteriochloroform): 50° δ 7.49 (d, Jg,9 = 5.1 Hz, 1H, H-8), 7.60-8.04 (m, 5H, ArH), 8.22 (d, J_{11.12} = 8.8 Hz, 1H, H-11), 9.21-9.36 (m, 1H, H-4).

Anal. Calcd. for C₁₇H₈ClNS₂: C, 62.66; H, 2.48; N, 4.30; S, 19.68. Found: C, 62.71; H, 2.50; N, 4.15; S, 19.47.

Benzo[h]thieno[2',3':4,5]thieno[2,3-c]quinoline (6).

A mixture of 0.43 g (1.32 mmoles) of 5, 74 mg (1.32 mmoles) of potassium hydroxide, 100 ml of methanol, 200 ml of benzene, and 0.1 g of 10% Pd-C was hydrogenated at atmospheric pressure and room temperature until the uptake of hydrogen ceased. The catalyst was removed by filtration and the filtrate was evaporated to dryness *in vacuo*. The solid was dissolved in 20 ml of chloroform and subjected to column chromatography, eluting with cyclohexane-benzene (2:1) mixture to recover 0.20 g of starting material 5 and to yield 0.10 g (0.34 mmole, 49%) of 6 as yellow prisms after recrystallization from cyclohexane, mp 212-213°; ir (potassium bromide): 3106, 3042 (aromatic CH stretching).

Anal. Calcd. for C₁₇H₉NS₂: C, 70.07; H, 3.11; N, 4.81. Found: C, 69.96; H, 3.25; N, 4.78.

3-Chloro-*N*-(2-naphthyl)thieno[3,2-*b*]thiophene-2-carboxamide (8).

A mixture of 2.25 g (10 mmoles) of 1 [6,7], 1.43 g (10 mmoles) of 2-naphthylamine (7), and 40 ml of benzene was heated under reflux for 4 hours. After cooling the solid was collected by filtration and recrystallized from benzene to give 2.73 g (7.94 mmoles, 79%) of 8 as colorless crystals, mp 199-201°; ir (potassium bromide): 3384 (NH stretching), 3096 (aromatic CH stretching), 1643 (C=O stretching); 1 H nmr (DMSO-d₆): δ 7.36-7.70 (m, 3H, ArH, including a doublet at δ 7.62 with J_{5,6} = 5.4 Hz attributed to H-6), 7.78-8.02 (m, 5H, ArH, including a doublet at δ 7.99 with J_{5,6} = 5.4 Hz assigned to H-5), 8.37 (d, J₁',₃' = 1.8 Hz, 1H, H-1'), 10.47 (br s, 1H, NH).

Anal. Calcd. for C₁₇H₁₀ClNOS₂: C, 59.38; H, 2.93; N, 4.07; S, 18.65. Found: C, 59.56; H, 3.10; N, 3.95; S, 18.43.

Benzo[f]thieno[2',3':4,5]thieno[2,3-c]quinolin-8(7H)-one (9).

A mixture of 0.2 g (0.58 mmole) of 8, 0.06 g of triethylamine in 500 ml of benzene was irradiated with a 450 watt Hanovia medium pressure mercury vapor lamp for 4 hours. A slow stream of air was passed through the solution during the course of the reaction. The solid was collected by filtration and washed with water to afford 0.18 g (0.58 mmole, 100%) of 9, mp, >280°; (ir potassium bromide): 3124 (NH stretching), 3088, 3068 (aromatic CH stretching), 1653 (C=O stretching); 1 H nmr (DMSO-d₆): 140° δ 7.46-7.80 (m, 4H, ArH, including a doublet at δ 7.65 with J_{10,11} = 5.4 Hz assigned to H-10), 7.91-8.10 (m, 3H, ArH, including a doublet at δ 7.94 with J_{10,11} = 5.4 Hz, for H-11), 8.96-9.07 (m, 1H, H-1). This compound was used in the next step without further purification because of low solubility.

8-Chlorobenzo[f]thieno[2',3':4,5]thieno[2,3-c]quinoline (10).

A mixture of 1.85 g (6.05 mmoles) of 9 and 45 ml of phosphorus oxychloride was heated at $110-120^{\circ}$ for 4 hours. After cooling the mixture was poured into 450 ml of ice-water with caution. The solid was collected by filtration and recrystallized from a benzene-cyclohexane mixture (1:1) to give 1.20 g (3.68 mmoles, 61%) of 10 as colorless crystals, mp 214-216°; ir (potassium bromide): 3098, 3080, 3044 (aromatic CH stretching); 1 H nmr (deuteriochloroform): δ 7.48 (d, $J_{10,11}$ = 5.4 Hz, 1H, H-10), 7.64-7.80 (m, 3H, ArH), 7.96-8.08 (m 3H, ArH), 9.25-9.35 (m, 1H, H-1).

Anal. Calcd. for C₁₇H₈ClNS₂: C, 62.66; H, 2.48; N, 4.30; S, 19.68. Found: 62.77; H, 2.72; N, 4.23; S, 19.49. Benzo[f]thieno[2',3':4,5]thieno[2,3-c]quinoline (11).

A mixture of 0.58 g (1.78 mmoles) of 10, 0.1 g of potassium hydroxide, 100 ml of methanol, 200 ml of benzene, and 0.1 g of 10% Pd-C was hydrogenated at atmospheric pressure and room temperature until the uptake of hydrogen ceased. The catalyst was removed by filtration and the filtrate was evaporated to dryness in vacuo. The residual solid was dissolved in 20 ml of chloroform and the solution was chromatographed on silica gel eluting with dichloromethane to recover 0.20 g of the starting material 10 and also to yield 0.20 g (0.69 mmole, 59%) of 11 as colorless crystals after recrystallization from cyclohexane, mp 170-172°; ir (potassium bromide): 3093, 3073, 3044 (aromatic CH stretching)

Anal. Calcd. for C₁₇H₉NS₂: C, 70.07; H, 3.11; N, 4.81. Found: C, 69.96; H, 3.14; N, 4.76

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